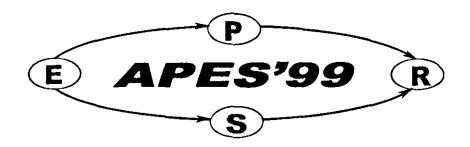
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PROCEEDINGS (ABSTRACT)





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CONDUCTION ESR AND $\alpha \leftrightarrow \beta$ PHASE TRANSFORMATIONS IN GRAPHITE INTERCALATION COMPOUNDS WITH NITRIC ACID

Nikolay M. MISHCHENKO and Albert M. ZIATDINOV

Institute of Chemistry, Far Eastern Branch of the Russian Academy of Sciences, 690022, Vladivostok, RUSSIA. E-mail: chemi@online.ru

The graphite intercalation compound (GIC) with nitric acid exists in two forms. When the ordinary form, α -C_{5n}HNO₃, is exposed to air or N₂ gas for an extended time period, the HNO₃ molecules, which stand essentially perpendicular to the graphite planes, reorient to lie nearly parallel, yielding the more dilute β -C_{8n}HNO₃ residue compound. Because synthesis of the residue compound is difficult and time consuming, most studies of HNO₃-GIC's have been confined to the α -type. In present work the results of Conduction ESR (CESR) and electroconductivity studies of $\alpha\square\beta$ phase transformations in HNO₃-GIC's are presented.

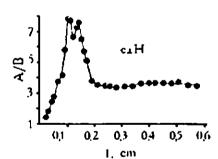


Figure 1. CESR line asymmetry parameter A/B (1) of β -C₂₀HNO₃ vers. sample width (/).

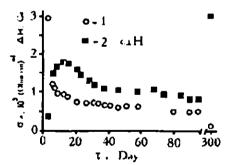


Figure 2. σ_e (1) and ΔH (2) of α -C₁₀HNO₃ vers exposure time instr.

At $\alpha\Box\beta$ phase transformation of $C_{10}HNO_3$ the electroconductivity in basal plane (σ_a) monotonically decreases from 2.9 • 10⁵ (Ω • cm)⁻¹ to 0.1 • 10⁵ (Ω • cm)⁻¹ (Fig. 2). Herewith, changing a CESR linewidth does not correlated with change σ_a (Fig. 2). At reintercalation of $\beta C_{24}HNO_3$ the value of σ_a - conductivity does not change. The value of electroconductivity along the c-axis (σ_c) practically does not depend on HNO₃-GIC's modification. In all studied GIC phases at decreasing of temperature a CESR linewidth increases together with σ_a .

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